Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

Bis{(1-methylimidazol-2-ylmethyl)[2-(2-pyridyl)ethyl]amine-κ^3N,N',N''}zinc(II) bis(hexafluoridophosphate)

Ai-Zhi Wu[a]* and Seik Weng Ng[b]

[a]School of Chinese Materia Medica, Guangzhou University of Chinese Medicine, Guangzhou 510006, People's Republic of China, and [b]Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia
Correspondence e-mail: wuaizhi@gzhtcm.edu.cn

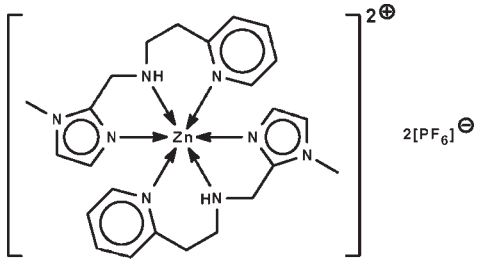
Received 19 October 2009; accepted 26 October 2009

Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.005 Å; R factor = 0.052; wR factor = 0.147; data-to-parameter ratio = 16.7.

Two tridentate *N*-heterocyclic ligands chelate the Zn^{II} atom in the title compound, $[Zn(C_{12}H_{16}N_4)_2](PF_6)_2$, conferring a *fac*-octahedral geometry. The Zn^{II} atom lies on a center of inversion. The cation is linked to the anion by an N—H···F hydrogen bond.

Related literature

No crystal structure studies of metal complexes with the *N*-heterocyclic ligand have been reported. For the synthesis of the ligand, see: Greatti *et al.* (2008).



Experimental

Crystal data

$[Zn(C_{12}H_{16}N_4)_2](PF_6)_2$
M_r = 787.89
Orthorhombic, *Pbca*
a = 13.3147 (5) Å
b = 11.8147 (4) Å
c = 20.3359 (6) Å
V = 3199.0 (2) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.97 mm^{-1}
T = 100 K
0.40 × 0.38 × 0.35 mm

Data collection

Bruker APEXII diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.698, T_{max} = 0.728
18287 measured reflections
3655 independent reflections
2740 reflections with $I > 2\sigma(I)$
R_{int} = 0.048

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.052
$wR(F^2)$ = 0.147
S = 1.04
3655 reflections
219 parameters
1 restraint
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 1.49 e Å^{-3}
$\Delta\rho_{min}$ = −0.71 e Å^{-3}

Table 1
Selected bond lengths (Å).

Zn1—N1	2.188 (3)	Zn1—N4	2.298 (3)
Zn1—N2	2.095 (3)		

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···F1	0.88 (1)	2.21 (1)	3.083 (4)	179 (3)

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

We thank Guangzhou University of Chinese Medicine and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2647).

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supplementary materials

Acta Cryst. (2009). E**65**, m1479 [**doi:10.1107/S1600536809044419**]

Bis{(1-methylimidazol-2-ylmethyl)[2-(2-pyridyl)ethyl]amine-κ^3N,N',N''}zinc(II) bis(hexafluoridophosphate)

A.-Z. Wu and S. W. Ng

Experimental

(1-Methylimidazol-2-ylmethyl)(pyridin-2-ylethyl)amine was synthesized according to a literature method (Greatti *et al.*, 2008).

The ligand (1 mmol, 0.26 g) dissolved in methanol (5 ml) was reacted with zinc hexafluorophosphat (1 mmol, 0.36 g) dissolved in water (5 ml). The mixture was filtered and the solution set aside for the growth of colorless block-shaped crystals that formed after several days in 60% yield. CH&N elemental analysis. Found: C 36.24, H 4.07, N 13.36%; calculated: C 36.55, H 4.06, N 14.22%.

Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.99 Å) and were included in the refinement in the riding model approximation, with *U*(H) set to 1.2 to 1.5*U*(C).

The imino H-atom was located in a difference Fourier map, and was refined with a distance restraint of N–H 0.88±0.01 Å; its temperature factor was freely refined.

The final difference Fourier map had a peak in the vicinity of Zn1 but was otherwise featureless.

Figures

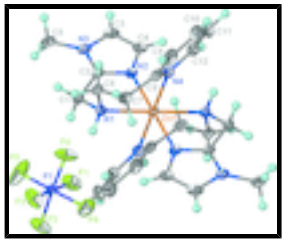


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of $[Zn(C_{12}H_{16}N_4)_2]$ 2$[PF_6]$ at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

Bis{(1-methylimidazol-2-ylmethyl)[2-(2-pyridyl)ethyl]amine- κ^3N,N',N''}zinc(II) bis(hexafluoridophosphate)

Crystal data

$[Zn(C_{12}H_{16}N_4)_2](PF_6)_2$	F_{000} = 1600
M_r = 787.89	D_x = 1.636 Mg m^{-3}
Orthorhombic, *Pbca*	Mo *K*α radiation, λ = 0.71073 Å
Hall symbol: -P 2ac 2ab	Cell parameters from 4790 reflections

a = 13.3147 (5) Å	θ = 2.5–28.2°
b = 11.8147 (4) Å	μ = 0.97 mm^{-1}
c = 20.3359 (6) Å	T = 100 K
V = 3199.0 (2) Å^3	Block, colorless
Z = 4	0.40 × 0.38 × 0.35 mm

Data collection

Bruker APEXII diffractometer	3655 independent reflections
Radiation source: fine-focus sealed tube	2740 reflections with $I > 2\sigma(I)$
Monochromator: graphite	R_{int} = 0.048
T = 100 K	θ_{max} = 27.5°
ω scans	θ_{min} = 2.0°
Absorption correction: Multi-scan (SADABS; Sheldrick, 1996)	h = −10→17
T_{min} = 0.698, T_{max} = 0.728	k = −13→15
18287 measured reflections	l = −22→26

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)]$ = 0.052	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2)$ = 0.147	$w = 1/[\sigma^2(F_o^2) + (0.0755P)^2 + 5.7073P]$ where $P = (F_o^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{max}$ = 0.001
3655 reflections	$\Delta\rho_{max}$ = 1.49 e Å^{-3}
219 parameters	$\Delta\rho_{min}$ = −0.71 e Å^{-3}
1 restraint	Extinction correction: none
Primary atom site location: structure-invariant direct methods	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
Zn1	0.5000	0.5000	0.5000	0.02098 (16)
P1	0.38657 (6)	0.83958 (7)	0.68434 (4)	0.0251 (2)
F1	0.47058 (17)	0.7507 (2)	0.66130 (14)	0.0524 (7)
F2	0.47025 (19)	0.9039 (2)	0.72513 (13)	0.0563 (7)
F3	0.30477 (18)	0.9306 (2)	0.70513 (12)	0.0540 (7)
F4	0.30581 (18)	0.7775 (2)	0.63902 (15)	0.0586 (7)
F5	0.3608 (2)	0.7641 (3)	0.74431 (16)	0.0840 (11)
F6	0.4135 (2)	0.9157 (2)	0.62167 (13)	0.0588 (7)
N1	0.40722 (19)	0.5353 (2)	0.58640 (13)	0.0231 (6)
H1	0.424 (3)	0.5969 (18)	0.6075 (15)	0.027 (9)*

N2	0.48361 (19)	0.3368 (2)	0.53888 (14)	0.0248 (6)
N3	0.4460 (2)	0.2326 (2)	0.62505 (14)	0.0282 (6)
N4	0.3548 (2)	0.4794 (2)	0.44007 (14)	0.0267 (6)
C1	0.4281 (3)	0.4463 (3)	0.63589 (16)	0.0274 (7)
H1A	0.3690	0.4358	0.6648	0.033*
H1B	0.4861	0.4685	0.6635	0.033*
C2	0.4510 (2)	0.3384 (3)	0.60004 (16)	0.0243 (7)
C3	0.4784 (3)	0.1600 (3)	0.57608 (19)	0.0326 (8)
H3	0.4838	0.0800	0.5790	0.039*
C4	0.5009 (2)	0.2245 (3)	0.5236 (2)	0.0294 (7)
H4	0.5248	0.1972	0.4825	0.035*
C5	0.4136 (3)	0.1995 (3)	0.69075 (18)	0.0372 (9)
H5A	0.3915	0.2667	0.7150	0.056*
H5B	0.4697	0.1638	0.7141	0.056*
H5C	0.3578	0.1457	0.6874	0.056*
C6	0.2980 (2)	0.5415 (3)	0.57167 (17)	0.0254 (7)
H6A	0.2621	0.5715	0.6106	0.031*
H6B	0.2724	0.4643	0.5629	0.031*
C7	0.2758 (2)	0.6170 (3)	0.51236 (17)	0.0292 (7)
H7A	0.2092	0.6529	0.5181	0.035*
H7B	0.3267	0.6778	0.5099	0.035*
C8	0.2766 (2)	0.5506 (3)	0.44925 (17)	0.0277 (7)
C9	0.1977 (3)	0.5560 (4)	0.40485 (19)	0.0391 (9)
H9	0.1448	0.6087	0.4116	0.047*
C10	0.1957 (3)	0.4854 (4)	0.3511 (2)	0.0498 (11)
H10	0.1425	0.4895	0.3201	0.060*
C11	0.2727 (3)	0.4086 (4)	0.34328 (19)	0.0449 (10)
H11	0.2724	0.3564	0.3078	0.054*
C12	0.3505 (3)	0.4095 (3)	0.38849 (17)	0.0330 (8)
H12	0.4040	0.3573	0.3825	0.040*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.0169 (3)	0.0213 (3)	0.0247 (3)	−0.00221 (18)	0.00329 (19)	0.00056 (19)
P1	0.0221 (4)	0.0233 (4)	0.0299 (5)	0.0017 (3)	−0.0017 (3)	−0.0007 (3)
F1	0.0394 (12)	0.0431 (13)	0.0747 (18)	0.0150 (11)	−0.0061 (12)	−0.0213 (12)
F2	0.0428 (13)	0.0550 (15)	0.0710 (18)	0.0110 (12)	−0.0256 (12)	−0.0275 (13)
F3	0.0449 (13)	0.0636 (16)	0.0534 (15)	0.0283 (12)	−0.0118 (11)	−0.0196 (12)
F4	0.0410 (13)	0.0396 (13)	0.095 (2)	−0.0040 (10)	−0.0220 (13)	−0.0184 (13)
F5	0.0588 (18)	0.106 (3)	0.087 (2)	0.0105 (17)	0.0142 (15)	0.069 (2)
F6	0.0594 (16)	0.0596 (16)	0.0573 (17)	−0.0066 (13)	−0.0012 (13)	0.0243 (13)
N1	0.0209 (12)	0.0230 (12)	0.0253 (15)	−0.0035 (10)	0.0010 (10)	−0.0006 (11)
N2	0.0218 (13)	0.0216 (12)	0.0309 (16)	−0.0020 (10)	0.0017 (11)	0.0006 (11)
N3	0.0248 (13)	0.0270 (14)	0.0328 (16)	−0.0074 (11)	−0.0075 (11)	0.0064 (12)
N4	0.0197 (13)	0.0328 (14)	0.0277 (15)	−0.0047 (11)	0.0012 (11)	0.0005 (11)
C1	0.0271 (15)	0.0277 (16)	0.0273 (18)	−0.0017 (13)	0.0014 (13)	0.0013 (13)
C2	0.0196 (14)	0.0218 (15)	0.0317 (18)	−0.0034 (12)	−0.0024 (12)	0.0040 (13)

C3	0.0272 (16)	0.0223 (15)	0.048 (2)	−0.0028 (13)	−0.0097 (15)	0.0008 (15)
C4	0.0221 (15)	0.0250 (16)	0.041 (2)	−0.0027 (12)	−0.0029 (14)	−0.0050 (14)
C5	0.043 (2)	0.0341 (18)	0.035 (2)	−0.0167 (16)	−0.0106 (16)	0.0121 (15)
C6	0.0171 (14)	0.0282 (15)	0.0310 (18)	−0.0024 (12)	0.0061 (12)	−0.0010 (13)
C7	0.0164 (14)	0.0285 (16)	0.043 (2)	0.0007 (12)	0.0027 (13)	0.0011 (14)
C8	0.0176 (14)	0.0335 (17)	0.0321 (19)	−0.0035 (12)	0.0036 (13)	0.0048 (14)
C9	0.0227 (17)	0.057 (2)	0.037 (2)	0.0016 (16)	−0.0007 (15)	0.0107 (18)
C10	0.0278 (19)	0.087 (3)	0.034 (2)	−0.006 (2)	−0.0050 (16)	0.003 (2)
C11	0.0316 (19)	0.076 (3)	0.027 (2)	−0.012 (2)	0.0021 (15)	−0.0128 (19)
C12	0.0235 (16)	0.044 (2)	0.031 (2)	−0.0061 (15)	0.0052 (13)	−0.0049 (15)

Geometric parameters (Å, °)

Zn1—N1	2.188 (3)	C1—H1A	0.9900
Zn1—N1^{i}	2.188 (3)	C1—H1B	0.9900
Zn1—N2^{i}	2.095 (3)	C3—C4	1.345 (5)
Zn1—N2	2.095 (3)	C3—H3	0.9500
Zn1—N4	2.298 (3)	C4—H4	0.9500
Zn1—N4^{i}	2.298 (3)	C5—H5A	0.9800
P1—F5	1.549 (3)	C5—H5B	0.9800
P1—F2	1.583 (2)	C5—H5C	0.9800
P1—F3	1.588 (2)	C6—C7	1.529 (5)
P1—F4	1.595 (2)	C6—H6A	0.9900
P1—F6	1.601 (3)	C6—H6B	0.9900
P1—F1	1.605 (2)	C7—C8	1.504 (5)
N1—C1	1.482 (4)	C7—H7A	0.9900
N1—C6	1.486 (4)	C7—H7B	0.9900
N1—H1	0.875 (10)	C8—C9	1.386 (5)
N2—C2	1.317 (4)	C9—C10	1.375 (6)
N2—C4	1.382 (4)	C9—H9	0.9500
N3—C2	1.351 (4)	C10—C11	1.378 (6)
N3—C3	1.383 (5)	C10—H10	0.9500
N3—C5	1.457 (4)	C11—C12	1.385 (5)
N4—C12	1.336 (4)	C11—H11	0.9500
N4—C8	1.352 (4)	C12—H12	0.9500
C1—C2	1.500 (4)		
N2^{i}—Zn1—N2	180.00 (6)	C2—C1—H1A	110.1
N2^{i}—Zn1—N1	100.73 (10)	N1—C1—H1B	110.1
N2—Zn1—N1	79.27 (10)	C2—C1—H1B	110.1
N2^{i}—Zn1—N1^{i}	79.27 (10)	H1A—C1—H1B	108.4
N2—Zn1—N1^{i}	100.73 (10)	N2—C2—N3	111.0 (3)
N1—Zn1—N1^{i}	180.0	N2—C2—C1	122.5 (3)
N2^{i}—Zn1—N4	89.13 (10)	N3—C2—C1	126.4 (3)
N2—Zn1—N4	90.87 (10)	C4—C3—N3	106.8 (3)
N1—Zn1—N4	88.35 (10)	C4—C3—H3	126.6
N1^{i}—Zn1—N4	91.65 (9)	N3—C3—H3	126.6
N2^{i}—Zn1—N4^{i}	90.87 (10)	C3—C4—N2	109.2 (3)

N2—Zn1—N4^{i}	89.13 (10)	C3—C4—H4	125.4
N1—Zn1—N4^{i}	91.65 (9)	N2—C4—H4	125.4
N1^{i}—Zn1—N4^{i}	88.35 (10)	N3—C5—H5A	109.5
N4—Zn1—N4^{i}	180.0	N3—C5—H5B	109.5
F5—P1—F2	91.12 (18)	H5A—C5—H5B	109.5
F5—P1—F3	91.64 (17)	N3—C5—H5C	109.5
F2—P1—F3	91.04 (13)	H5A—C5—H5C	109.5
F5—P1—F4	92.36 (18)	H5B—C5—H5C	109.5
F2—P1—F4	176.28 (17)	N1—C6—C7	112.2 (2)
F3—P1—F4	90.16 (13)	N1—C6—H6A	109.2
F5—P1—F6	179.05 (19)	C7—C6—H6A	109.2
F2—P1—F6	89.42 (16)	N1—C6—H6B	109.2
F3—P1—F6	89.12 (15)	C7—C6—H6B	109.2
F4—P1—F6	87.08 (15)	H6A—C6—H6B	107.9
F5—P1—F1	90.41 (17)	C8—C7—C6	111.6 (3)
F2—P1—F1	88.66 (13)	C8—C7—H7A	109.3
F3—P1—F1	177.93 (16)	C6—C7—H7A	109.3
F4—P1—F1	90.02 (13)	C8—C7—H7B	109.3
F6—P1—F1	88.83 (15)	C6—C7—H7B	109.3
C1—N1—C6	110.8 (2)	H7A—C7—H7B	108.0
C1—N1—Zn1	107.68 (19)	N4—C8—C9	121.5 (3)
C6—N1—Zn1	113.6 (2)	N4—C8—C7	116.6 (3)
C1—N1—H1	102 (2)	C9—C8—C7	121.8 (3)
C6—N1—H1	108 (2)	C10—C9—C8	120.3 (4)
Zn1—N1—H1	114 (2)	C10—C9—H9	119.8
C2—N2—C4	106.3 (3)	C8—C9—H9	119.8
C2—N2—Zn1	112.2 (2)	C9—C10—C11	118.5 (4)
C4—N2—Zn1	141.4 (2)	C9—C10—H10	120.8
C2—N3—C3	106.7 (3)	C11—C10—H10	120.8
C2—N3—C5	127.4 (3)	C10—C11—C12	118.3 (4)
C3—N3—C5	125.9 (3)	C10—C11—H11	120.8
C12—N4—C8	117.4 (3)	C12—C11—H11	120.8
C12—N4—Zn1	121.2 (2)	N4—C12—C11	123.9 (4)
C8—N4—Zn1	120.6 (2)	N4—C12—H12	118.0
N1—C1—C2	108.1 (3)	C11—C12—H12	118.0
N1—C1—H1A	110.1		

N2^{i}—Zn1—N1—C1	−152.89 (19)	Zn1—N2—C2—C1	0.5 (4)
N2—Zn1—N1—C1	27.11 (19)	C3—N3—C2—N2	0.4 (4)
N4—Zn1—N1—C1	118.3 (2)	C5—N3—C2—N2	−180.0 (3)
N4^{i}—Zn1—N1—C1	−61.7 (2)	C3—N3—C2—C1	−176.9 (3)
N2^{i}—Zn1—N1—C6	84.0 (2)	C5—N3—C2—C1	2.7 (5)
N2—Zn1—N1—C6	−96.0 (2)	N1—C1—C2—N2	23.2 (4)
N4—Zn1—N1—C6	−4.8 (2)	N1—C1—C2—N3	−159.8 (3)
N4^{i}—Zn1—N1—C6	175.2 (2)	C2—N3—C3—C4	−0.4 (4)
N1—Zn1—N2—C2	−15.5 (2)	C5—N3—C3—C4	179.9 (3)
N1^{i}—Zn1—N2—C2	164.5 (2)	N3—C3—C4—N2	0.3 (4)
N4—Zn1—N2—C2	−103.7 (2)	C2—N2—C4—C3	−0.1 (4)

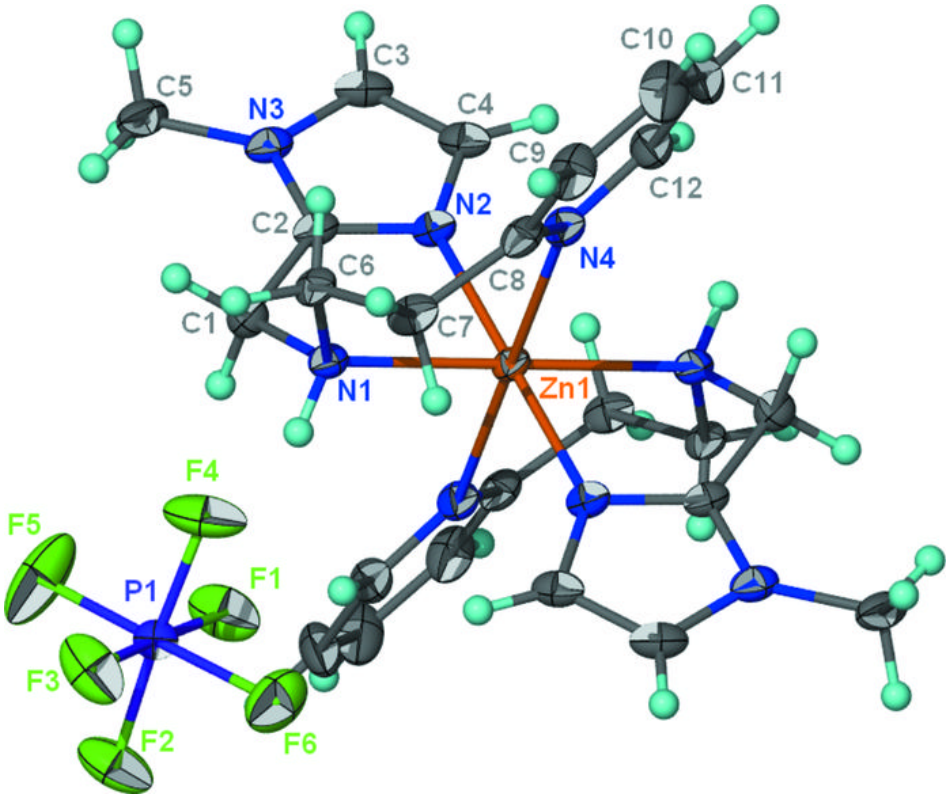
N4^{i}—Zn1—N2—C2	76.3 (2)	Zn1—N2—C4—C3	175.0 (3)
N1—Zn1—N2—C4	169.6 (4)	C1—N1—C6—C7	−169.8 (3)
N1^{i}—Zn1—N2—C4	−10.4 (4)	Zn1—N1—C6—C7	−48.4 (3)
N4—Zn1—N2—C4	81.4 (4)	N1—C6—C7—C8	90.1 (3)
N4^{i}—Zn1—N2—C4	−98.6 (4)	C12—N4—C8—C9	−3.8 (5)
N2^{i}—Zn1—N4—C12	109.5 (3)	Zn1—N4—C8—C9	165.9 (3)
N2—Zn1—N4—C12	−70.5 (3)	C12—N4—C8—C7	171.8 (3)
N1—Zn1—N4—C12	−149.8 (3)	Zn1—N4—C8—C7	−18.5 (4)
N1^{i}—Zn1—N4—C12	30.2 (3)	C6—C7—C8—N4	−47.0 (4)
N2^{i}—Zn1—N4—C8	−59.9 (2)	C6—C7—C8—C9	128.7 (3)
N2—Zn1—N4—C8	120.1 (2)	N4—C8—C9—C10	2.3 (6)
N1—Zn1—N4—C8	40.9 (2)	C7—C8—C9—C10	−173.1 (3)
N1^{i}—Zn1—N4—C8	−139.1 (2)	C8—C9—C10—C11	1.0 (6)
C6—N1—C1—C2	92.0 (3)	C9—C10—C11—C12	−2.6 (6)
Zn1—N1—C1—C2	−32.7 (3)	C8—N4—C12—C11	2.2 (5)
C4—N2—C2—N3	−0.2 (3)	Zn1—N4—C12—C11	−167.5 (3)
Zn1—N2—C2—N3	−176.9 (2)	C10—C11—C12—N4	1.1 (6)
C4—N2—C2—C1	177.2 (3)		

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*+1.

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···F1	0.88 (1)	2.21 (1)	3.083 (4)	179 (3)

Fig. 1


C10
C11
C3
C5
N3
C4
C9
C12
N2
C2
N4
C6
C8
C1
C7
N1
Zn1
F4
F5
F1
P1
F3
F2
F6